

AMENDMENT TO SPECIFICATION

Please amend the paragraph bridging pages 4 and 5 as follows.

1 The halide-donating compound for the co-electrolyte is preferably a  
2 molecule capable of reacting with a Lewis acid, preferably a strong Lewis acid,  
3 by donating a halogen atom. Preferably the halide-donating compound is  $\text{RSO}_2\text{X}$ ,  
4  $\text{RP}(\text{O})\text{X}_2$ , or combinations thereof where R is  $-\text{CX}'_3$ ,  $-\text{N}=\text{PX}_3$ ,  $-(\text{CX}_2)_n\text{CX}_3$ , or  
5 combinations of two or more thereof, X is a halogen atom, X' is hydrogen, a  
6 halogen, or combinations thereof, and  $n = 3-7$ . Most preferably X is a chlorine  
7 atom. More preferably, the halide-donating compound is a largely or  
8 substantially inorganic compound. Especially preferred halide-donating  
9 compounds include, but are not limited to, methane sulfonyl chloride,  
10 trichlorophosphazophosphoryl chloride ( $\text{Cl}_3\text{P}=\text{NP}(\text{O})\text{Cl}_2$ ),  
11 ~~trichlorophosphazosulfonyl~~ trichlorophosphazosulfonyl chloride  
12 ( $\text{Cl}_3\text{P}=\text{NSO}_2\text{Cl}$ ), and combinations of two or more thereof. These halide-donating  
13 compounds can be produced by any means known to one skilled in the art such as,  
14 for example, the means disclosed in J. Electrochem. Soc. Vol. 143 No. 11, pages  
15 3548-3554, November 1996.